

chem 5390

# ***Advanced X-ray Analysis***



## **LECTURE 16**

**Dr. Teresa D. Golden**  
**University of North Texas**  
**Department of Chemistry**

# Application of Diffraction Data

**XRD can be used for:**

- Bravais lattice determination – phase determination (crystalline phases and orientation)
- Lattice parameter determination
- Determination of solvus line in phase diagrams (order-disorder transformation)
- Long range order
- Crystallite size and Strain
- Temperature factor – thermal diffuse scattering (thermal expansion)
- Thickness measurements of thin films and multilayers

# Application of Diffraction Data

## XRD can be used for:

- Bravais lattice determination – phase determination (crystalline phases and orientation)
- Lattice parameter determination
- Determination of solvus line in phase diagrams (order-disorder transformation)**
- Long range order
- Crystallite size and Strain
- Temperature factor – thermal diffuse scattering (thermal expansion)
- Thickness measurements of thin films and multilayers

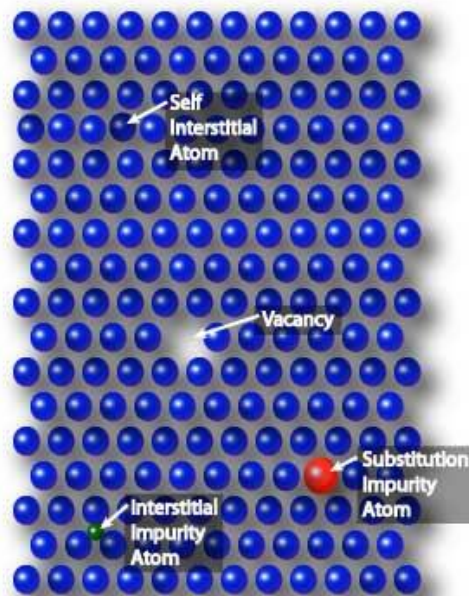
# Application of Diffraction Data

## Vectors and Planes

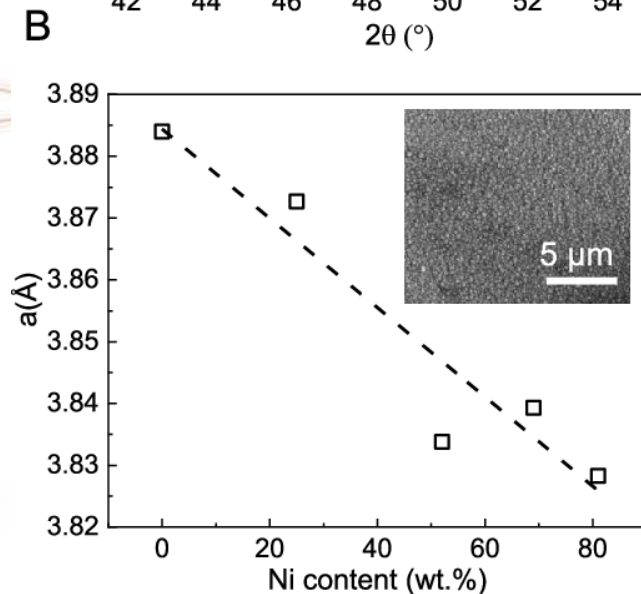
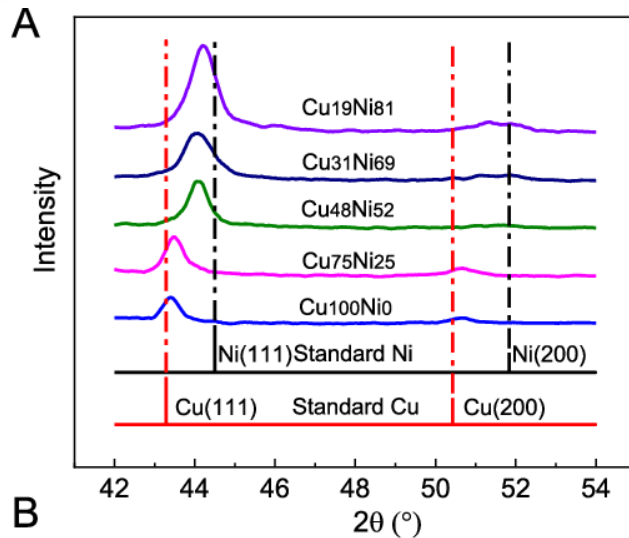
### Crystal Defects

Point defects – substitutional or interstitial impurities.

Vegard's law – volume of unit cells in a substitutional solid solution is linearly proportional to the fraction of sites substituted.



# Application of Diffraction Data



## Vegard's law

In crystallography, Vegard's law is an empirical finding resembling the rule of mixtures. In 1921, Lars Vegard discovered that the lattice parameter of a solid solution of two constituents is approximately a weighted mean of the two constituents' lattice parameters at the same temperature.

# Application of Diffraction Data

## Phase Analysis - Vegard's Law

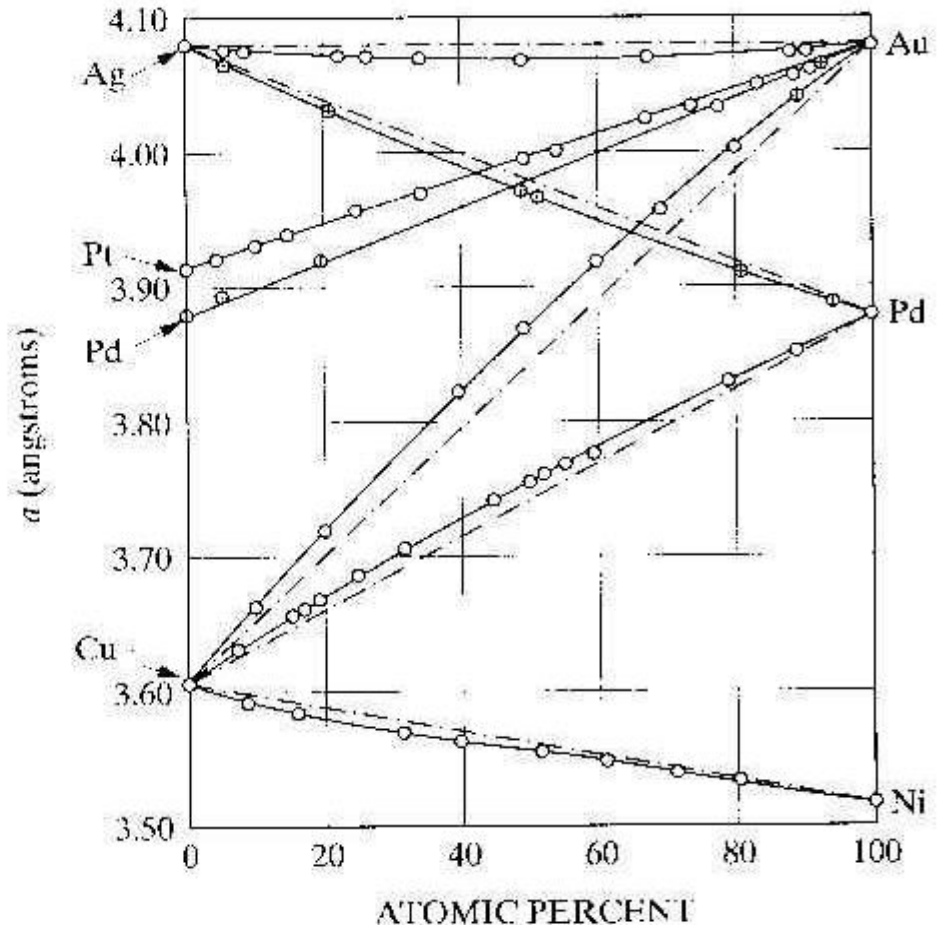
$$d_{hkl} = \sqrt{\frac{a^2}{h^2 + k^2 + l^2}}$$

### Ex) Au-Pd

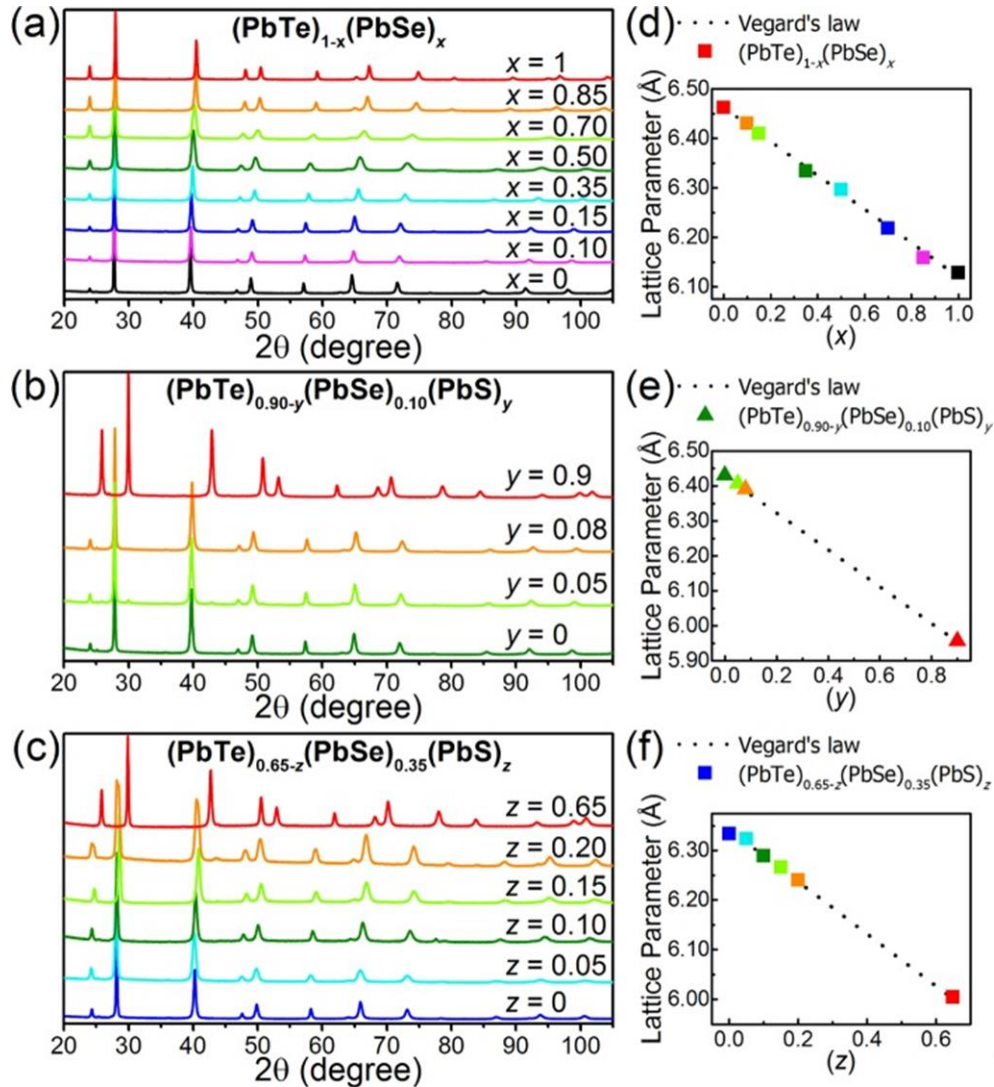
- To create the plot on the right
  - Using the crystal structure of the alloy calculate "a" for each metal
  - Draw a straight line between them as shown on the chart to the left.
- To calculate the composition
  - Calculate "a" from d-spacings
  - "a" will be an atomic weighted fraction of "a" of the two metal

Figure 11-6 Lattice parameters of some continuous solid solutions. Dot-dash lines indicate Vegard's law. Barrett [1.7].

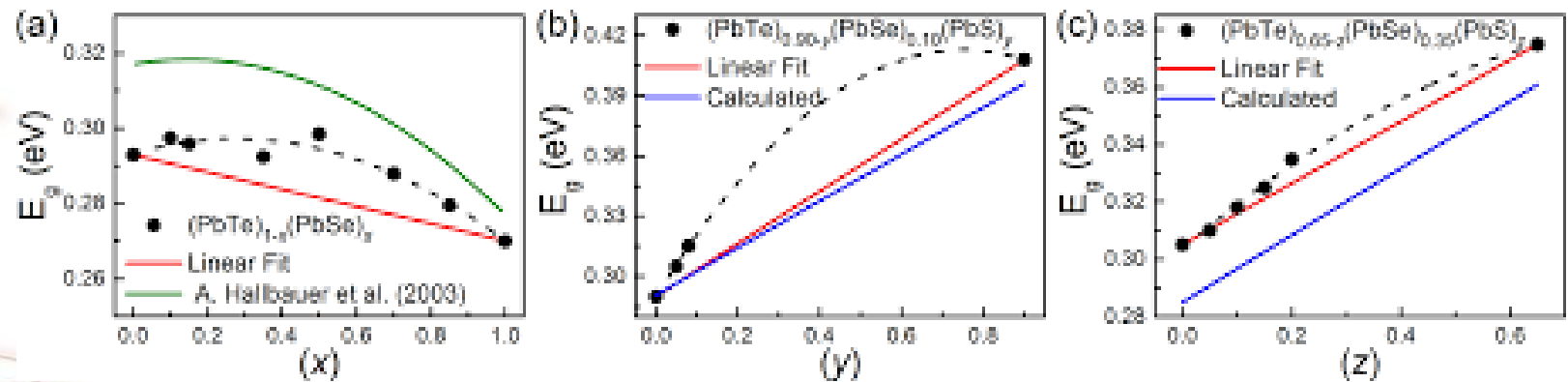
Good for alloys with continuous solid solutions



# Application of Diffraction Data



# Application of Diffraction Data





# Application of Diffraction Data

## Measuring Changes In Phase Fraction

Using  $I/I_{cor}$

$$\frac{I_{\alpha\text{-exp}}(hkl)}{I_{\beta\text{-exp}}(HKL)} = \frac{I_{\alpha}(hkl) \times I/I_{cor\beta}(hkl) \times w_{\beta}}{I/I_{cor\alpha}(hkl) \times I_{\beta}(hkl) \times w_{\alpha}}$$

–Where

- $\frac{I}{I_{cor}} = \frac{\text{Intensity of sample's 100\% peak}}{\text{Intensity of Corundum's 100\% peak}}$
- $\omega$  = weight fraction
- $I(hkl)$  = Reference's relative intensity
- $I_{exp}(hkl)$  = Experimental integrated intensity

# Application of Diffraction Data

## Measuring Changes In Phase Fraction

Using Direct Comparison Method

$$\frac{I_{\alpha\text{-exp}}(hkl)}{I_{\beta\text{-exp}}(HKL)} = \frac{R_{\alpha} \times v_{\alpha}}{R_{\beta} \times v_{\beta}}$$

$$R = \frac{1}{V} \left[ pCL_P [F_{hkl}]^2 \right]$$

$$I_{hkl} = I_0 pCL_P [F_{hkl}]^2$$

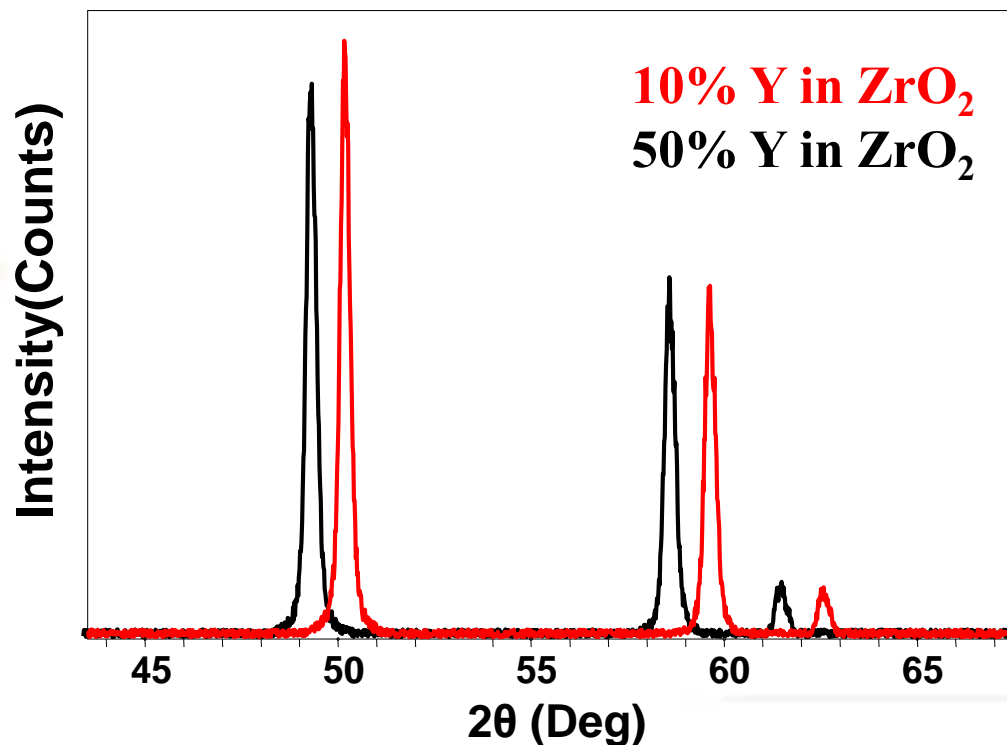
**Because this is already a complicated method, many choose to go ahead and use Rietveld Refinement**

# Application of Diffraction Data

## Measuring Changes In Phase Fraction

Substitutional doping can change bond distances, reflected by a change in unit cell lattice parameters

The change in peak intensity due to substitution of atoms with similar Z is much more subtle and may be insignificant



$$R(\text{Y}^{3+}) = 0.104\text{\AA}$$

$$R(\text{Zr}^{4+}) = 0.079\text{\AA}$$

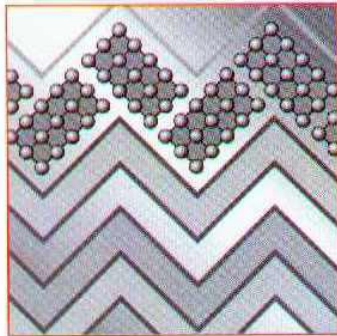
# Application of Diffraction Data

## XRD can be used for:

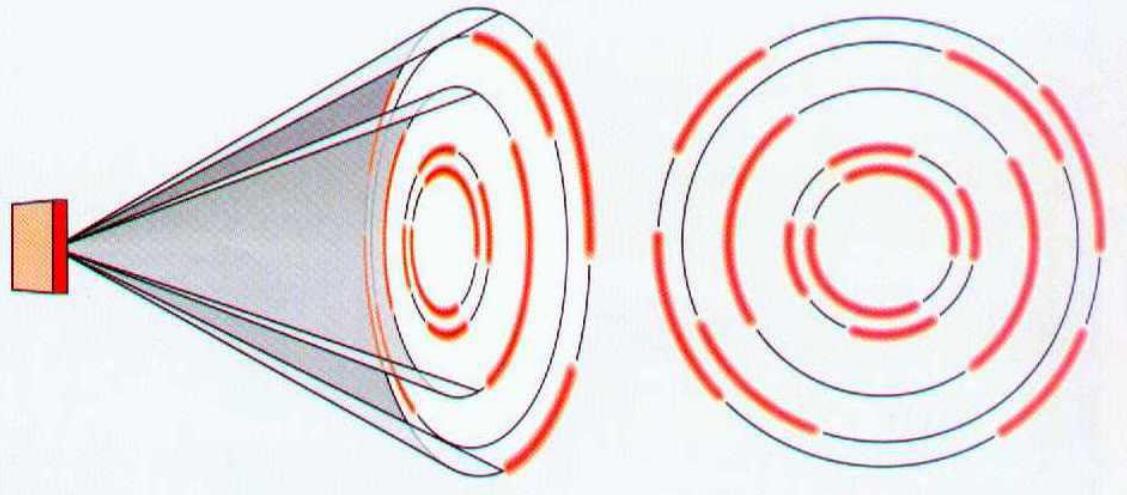
- Bravais lattice determination – phase determination (crystalline phases and orientation)
- Lattice parameter determination
- Determination of solvus line in phase diagrams (order-disorder transformation)
- Long range order (Texture Analysis)**
- Crystallite size and Strain
- Temperature factor – thermal diffuse scattering (thermal expansion)
- Thickness measurements of thin films and multilayers

# Diffraction Theory

## Texture analysis

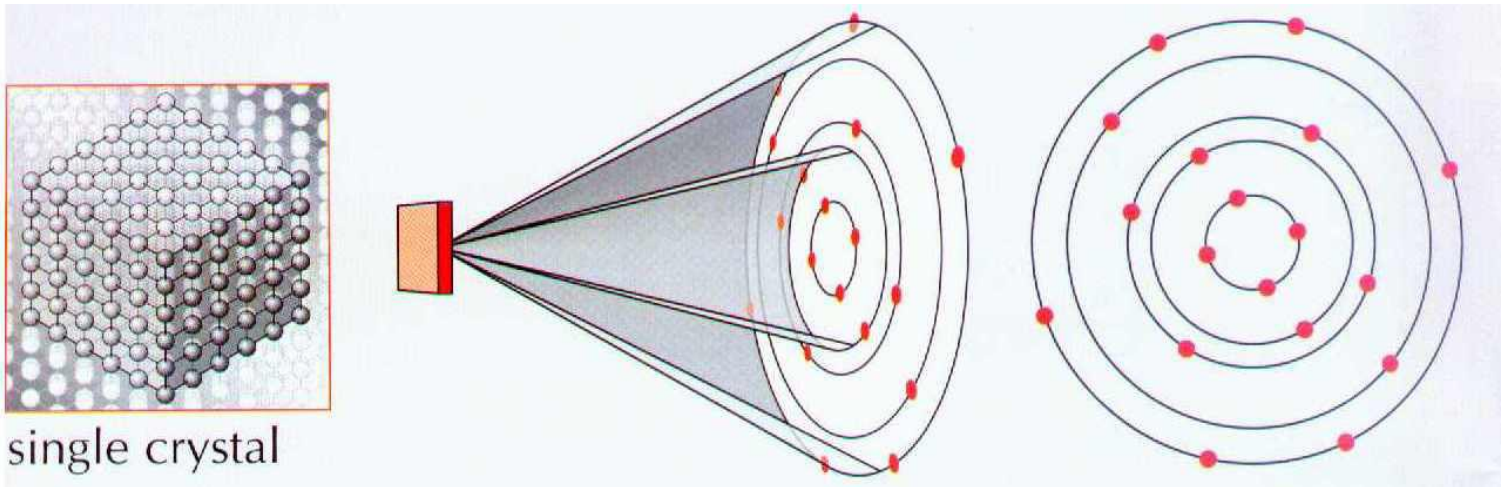


texture

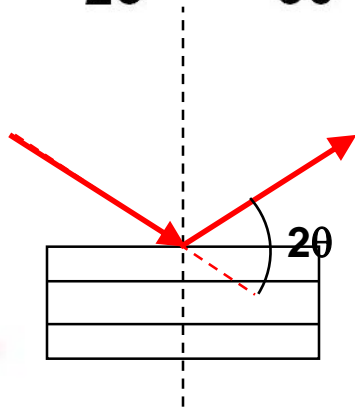
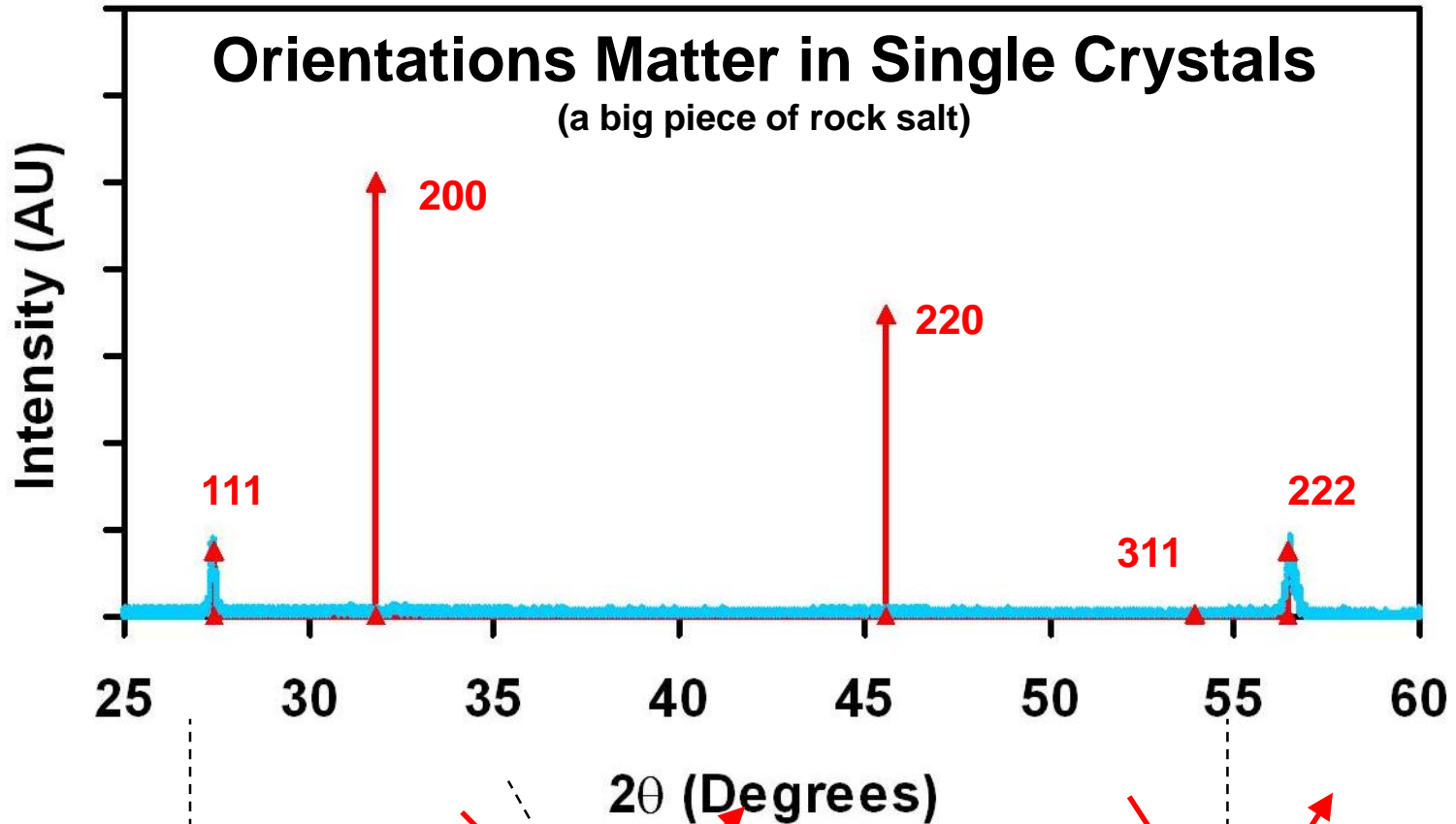


# Diffraction Theory

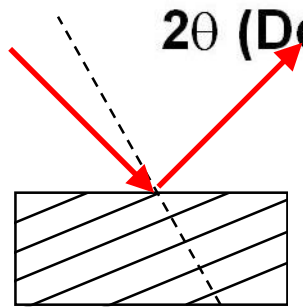
## Single Crystal



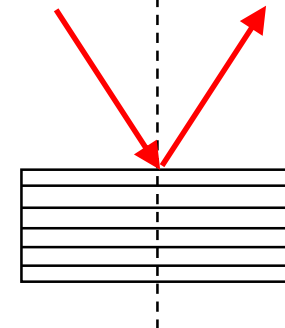
# Application of Diffraction Data



At  $27.42^\circ$   $2\theta$ , Bragg's law fulfilled for the (111) planes, producing a diffraction peak.



The (200) planes would diffract at  $31.82^\circ$   $2\theta$ ; however, they are not properly aligned to produce a diffraction peak



The (222) planes are parallel to the (111) planes.

# Application of Diffraction Data

## Texture

Grains within a polycrystalline are not completely randomly distributed

There can be clustering of grains about some particular orientation(s) to a certain degree

Inducing texture in powder samples - causes peak intensities to vary

Large crystallite sizes and non-random crystallite orientations both lead to peak intensity variation

The measured diffraction pattern will not agree with that expected from an ideal powder



# Application of Diffraction Data

Texture in samples is a common occurrence, and usually have more to do with how the samples are made.

- Plastically deformed metals (cold rolled, etc)
- Powders with particle shapes related to their crystal structure, particular planes form the faces
- Orientation in particular directions (Plates, needles, acicular, cubes, etc.)

# Application of Diffraction Data

Orientation in particular directions (Plates, needles, acicular, cubes, etc.)

## -Fiber Texture

A particular direction  $[uvw]$  for all grains is more or less parallel to the wire or fiber axis

e.g.  $[111]$  fiber texture in Al cold drawn wire

Double axis is also possible

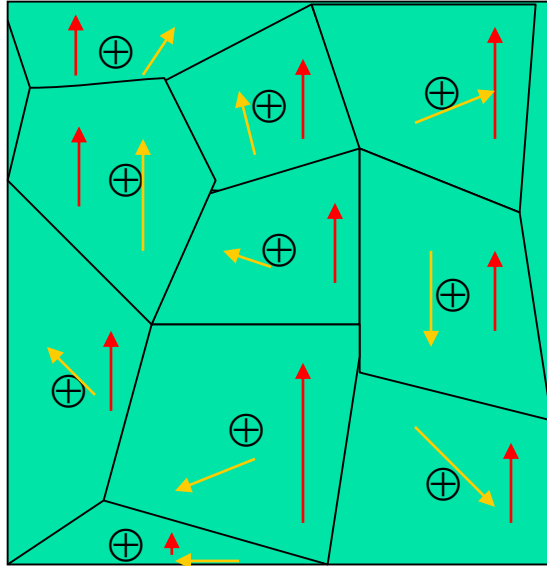
Example:  $[111]$  and  $[100]$  fiber textures in Cu wire

## -Sheet Texture

Most of the grains are oriented with a certain crystallographic plane  $(hkl)$  roughly parallel to the sheet surface and certain direction  $[uvw]$  parallel to the rolling direction

Notation:  $(hkl)[uvw]$

# Application of Diffraction Data



$\oplus$   $[uvw]$  i.e. perpendicular to the surface of all grains is parallel to a direction  $[uvw]$

**Orientation in particular directions (Plates, needles, acicular, cubes, etc.)**

Also, if the direction  $[u_1v_1w_1]$  is parallel for all regions, the structure is like a single crystal

However, the direction  $[u_1v_1w_1]$  is not aligned for all regions, the structure is like a mosaic structure, also called as **Mosaic Texture**

# Application of Diffraction Data

## Simple Means of Quantifying Texture

### Lotgering degree of orientation ( $f$ )

A comparison of the relative intensities of a particular family of (hkl) reflections to all observed reflections in a coupled  $2\theta$  powder x-ray diffraction (XRD) Spectrum

$f$  is specifically considered a measure of the “degree of orientation” and ranges from 0% to 100%

$p_0$  is  $p$  of a sample with a random crystallographic orientation.

$$f = \frac{p - p_0}{1 - p_0}$$

for (00l)

$$p = \frac{\sum I_{00l}}{\sum I_{00l} + \sum I_{\text{non-00l}}}$$

# Application of Diffraction Data

## Relative Texture Coefficient (RTC)

Relative peak intensities are analyzed by determining relative texture coefficients of the reflections present in the XRD pattern.

$$RTC_{hkl} = \frac{I_{p,hkl} / I^{\circ}_{p,hkl}}{\sum I_{p,hkl} / I^{\circ}_{p,hkl}} * 100\%$$

where  $I_{p,hkl}$  is the intensity of the peak for each sample at each (hkl) reflection used.

$I^{\circ}_{p,hkl}$  is the intensity of those reflections for the standard reference (PDF# 00-000-0000) in the database.

The standard provides a random orientation pattern to compare against any preferred orientation found in the samples.

# Application of Diffraction Data

Relative Texture Coefficient (RTC) Example: Golden et.al.

X-ray diffraction (XRD)  
of (a) pure Ni, (b) Ni CeO<sub>2</sub>, and  
(c) Ni Fc-CeO<sub>2</sub> coatings

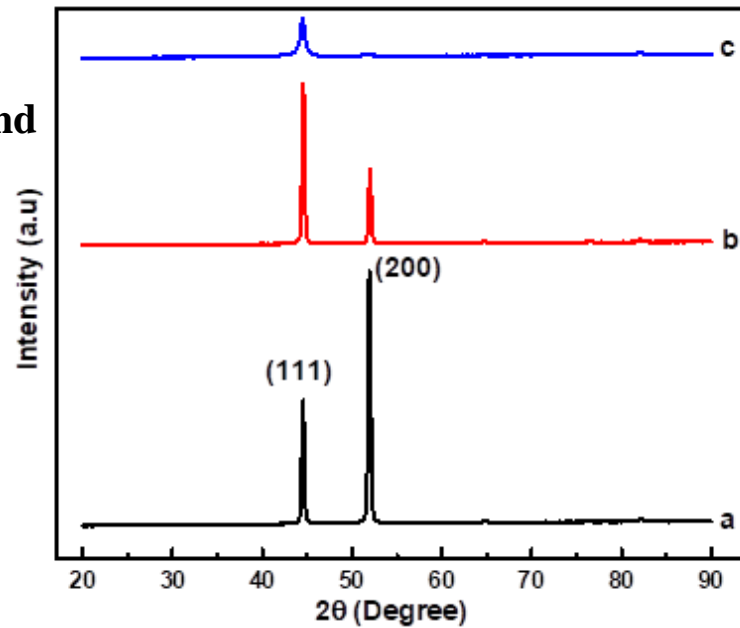
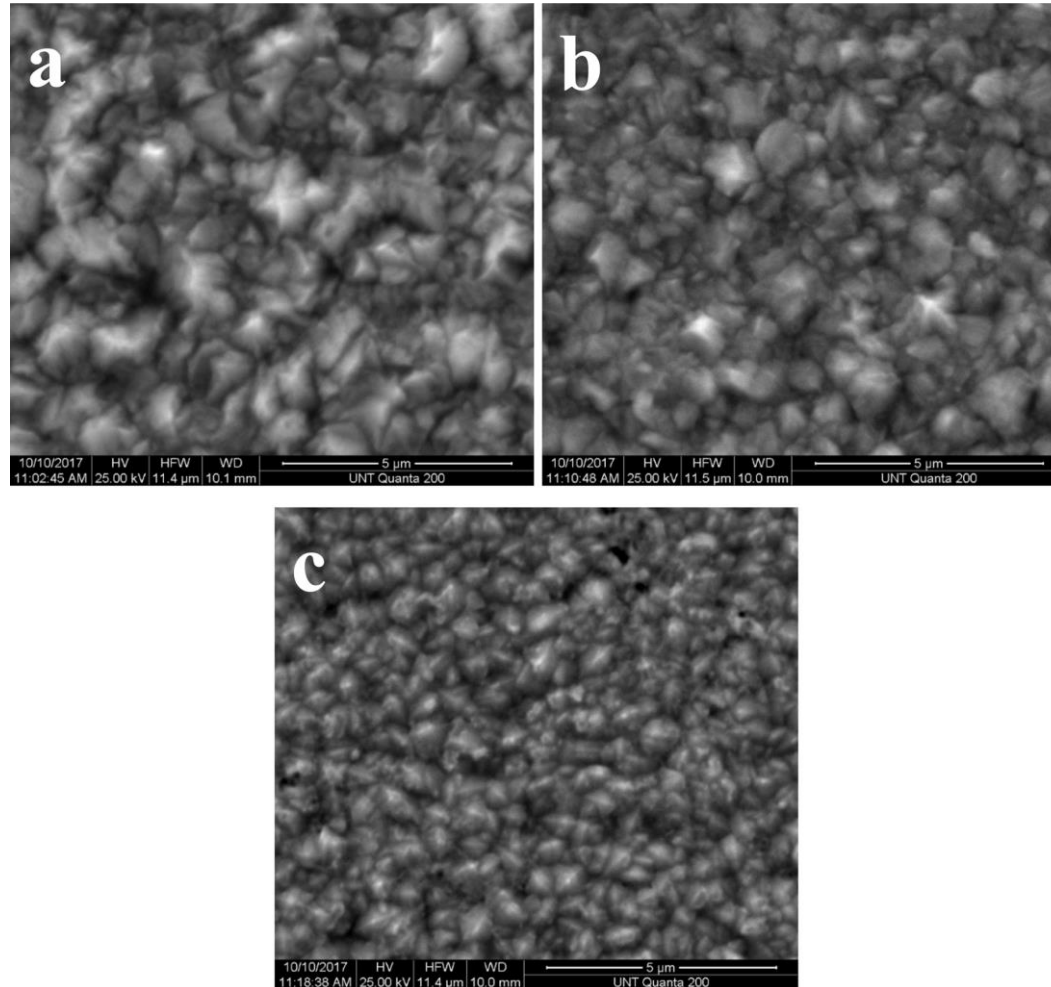


Table. XRD reflection intensities, relative texture coefficients, and crystallite sizes for Nickel, Ni CeO<sub>2</sub> and Ni Fc-CeO<sub>2</sub> electrodeposited coatings.

| Coating (n=3)          | I(f) <sub>111</sub> | I(f) <sub>200</sub> | I(f) <sub>220</sub> | RTC <sub>111</sub> | RTC <sub>200</sub> | RTC <sub>220</sub> | Crystallite size (nm) |
|------------------------|---------------------|---------------------|---------------------|--------------------|--------------------|--------------------|-----------------------|
| PDF#<br>00-004-0850    | 100                 | 42                  | 21                  | ---                | ---                | ---                | ---                   |
| Nickel                 | 49                  | 100                 | 3                   | 29.8               | 63.3               | 6.9                | 47.6 ± 5.0            |
| Ni CeO <sub>2</sub>    | 100                 | 51                  | 4                   | 50.7               | 41.9               | 6.7                | 43.3 ± 6.3            |
| Ni Fc-CeO <sub>2</sub> | 100                 | 11                  | 16                  | 63.1               | 14.6               | 22.3               | 32.0 ± 5.5            |

# Application of Diffraction Data

Relative Texture Coefficient (RTC) Example: Golden et.al.



# Application of Diffraction Data

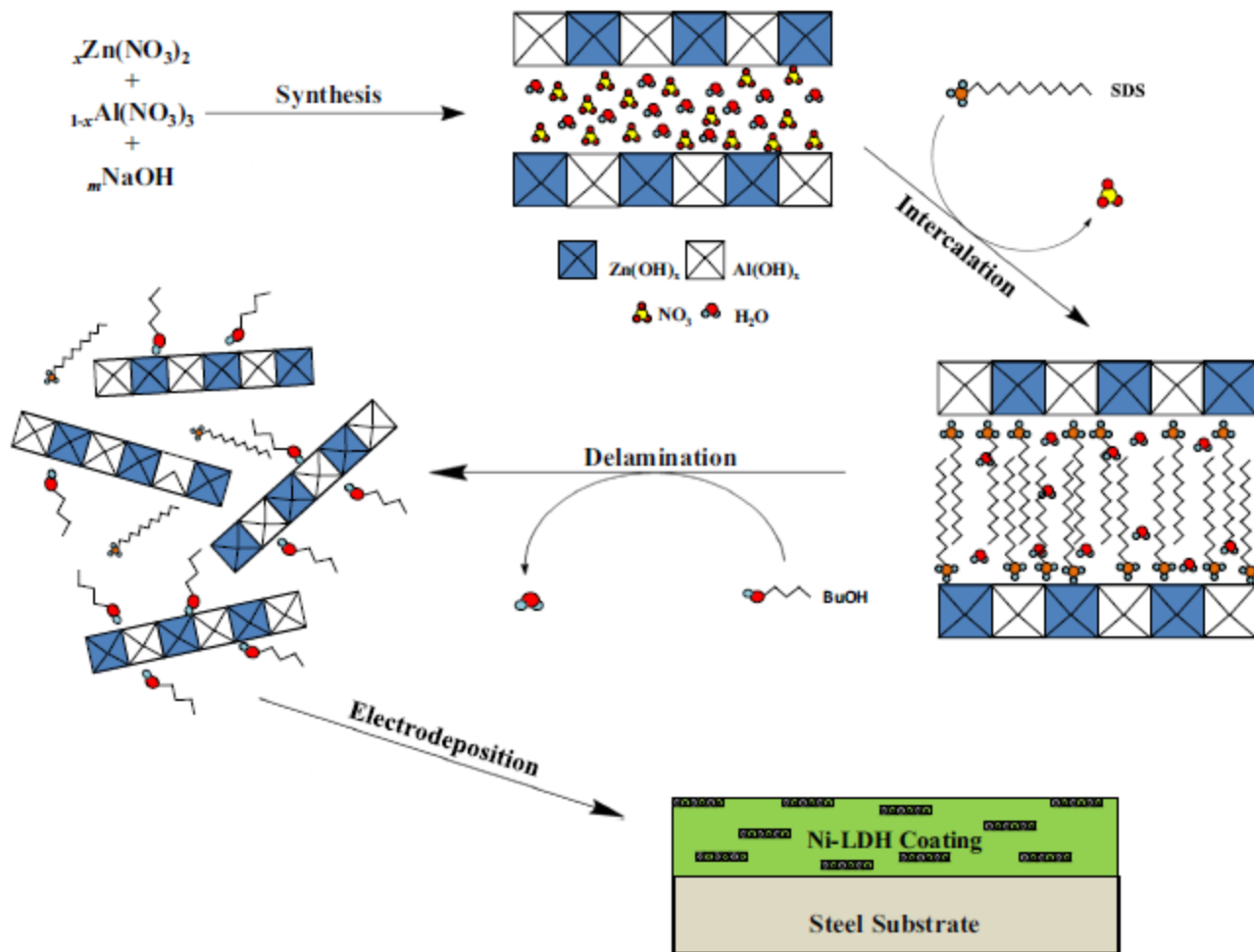


Fig. 1. Overview of the layered double hydroxide colloid synthesis and Ni-LDH metal matrix nanocomposite coating process.



# Application of Diffraction Data

Table 2

Crystallite size and strain measured from Williamson-Hall analysis and results of relative texture coefficient analysis of the XRD patterns of Ni and Ni-LDH coatings.

|     | Crystallite size (nm) | Strain         | RTC <sub>111</sub> | RTC <sub>200</sub> | RTC <sub>220</sub> | RTC <sub>311</sub> |
|-----|-----------------------|----------------|--------------------|--------------------|--------------------|--------------------|
| B1  | 37.54 ± 27            | 0.0228 ± 0.013 | 10.04 ± 5.5        | 5.84 ± 2.2         | 75.93 ± 9.0        | 8.19 ± 1.5         |
| B2  | 60.20 ± 21            | 0.0192 ± 0.001 | 9.86 ± 6.0         | 8.66 ± 4.6         | 72.36 ± 12.0       | 9.12 ± 2.0         |
| BL1 | 41.16 ± 2             | 0.0120 ± 0.007 | 6.25 ± 0.2         | 3.40 ± 1.0         | 84.94 ± 4.5        | 5.41 ± 3.3         |
| BL2 | 25.10 ± 6             | 0.0158 ± 0.017 | 8.22 ± 1.8         | 5.88 ± 3.3         | 78.91 ± 8.6        | 6.99 ± 3.7         |

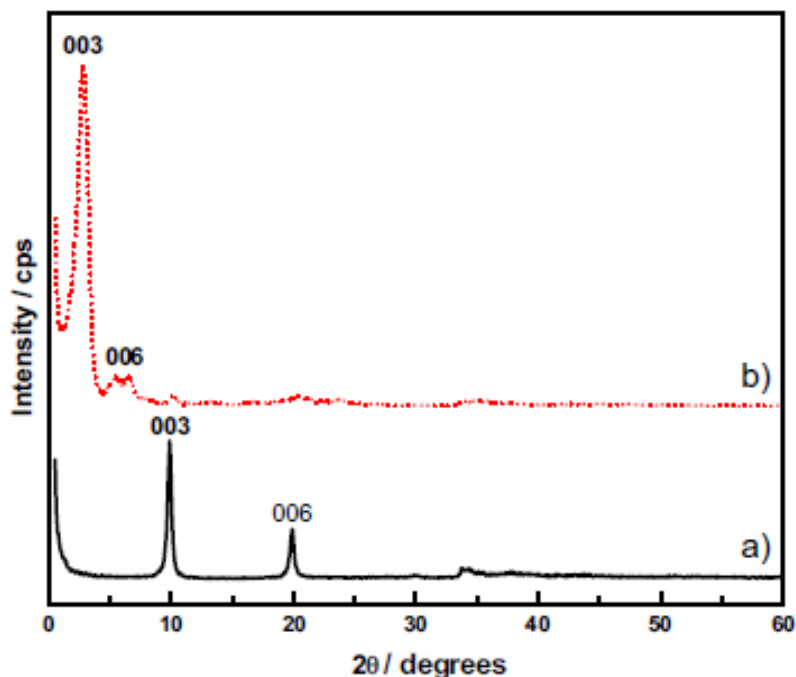


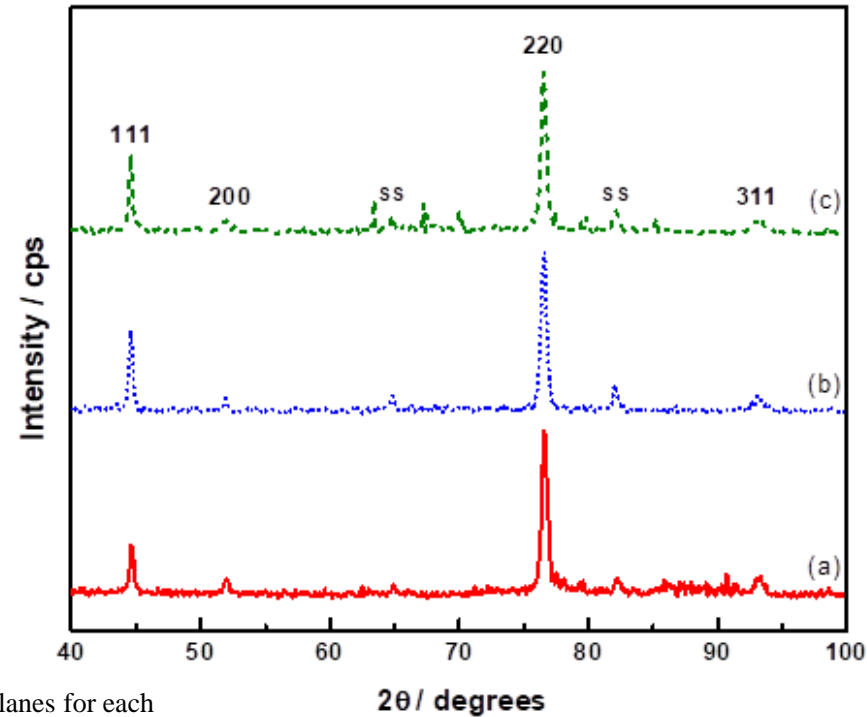
Fig. 3. Powder XRD patterns of a) LDH-NO<sub>3</sub> (solid line), and b) LDH-DS (dotted line) powders.

# Application of Diffraction Data

## Relative Texture Coefficient (RTC) Example: Golden et.al.

$$RTC_{hkl} = \frac{I_{p,hkl}/I_{p,hkl}^o}{\sum I_{p,hkl}/I_{p,hkl}^o} \times 100\%$$

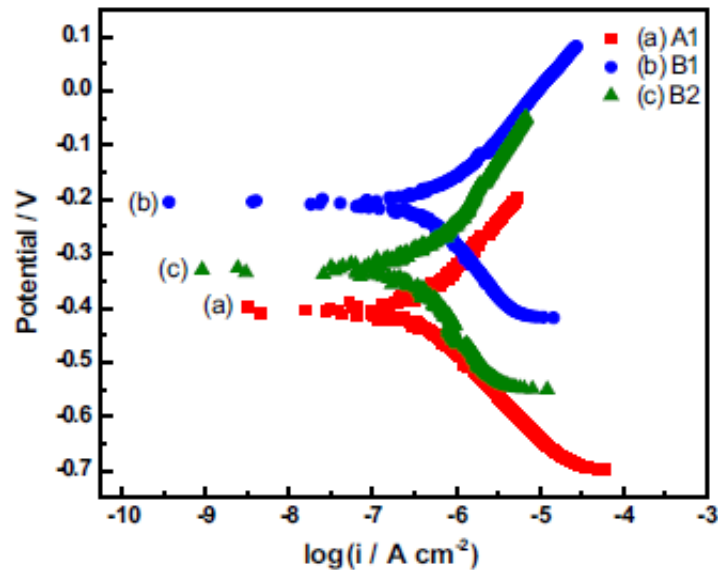
where  $I_{p,hkl}$  is the intensity of the peak for each sample at (111), (200), (220), and (311) reflections and  $I_{p,hkl}^o$  is the intensity of those reflections for the nickel standard reference (PDF# 00-004-0850)



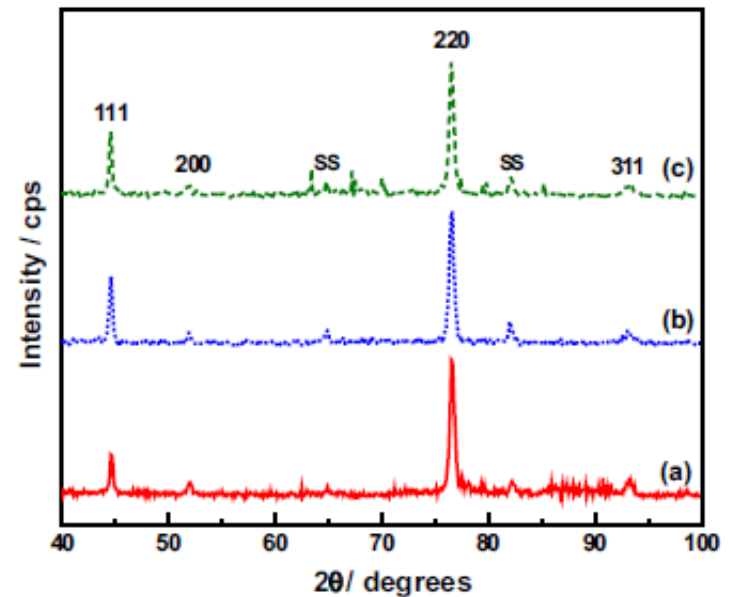
**Table 2:** Relative texture coefficients for (111), (200), (220), and (311) growth planes for each sample as measured with XRD.

| Sample | RTC <sub>111</sub> | RTC <sub>200</sub> | RTC <sub>220</sub> | RTC <sub>311</sub> |
|--------|--------------------|--------------------|--------------------|--------------------|
| A1     | 5.26               | 4.85               | 72.32              | 17.66              |
| B1     | 8.09               | 4.32               | 71.84              | 15.74              |
| B2     | 7.80               | 4.27               | 72.66              | 15.26              |

# Application of Diffraction Data



**Fig. 7** Tafel polarization curves for each coating *a* A1 (squares), *b* B1 (circles), and *c* B2 (triangles) run in 3.5% NaCl after 14 days immersion in 3.5% NaCl



**Fig. 2** X-ray diffraction patterns of each nickel coating *a* A1 (solid line), *b* B1 (dotted line), and *c* B2 (dashed line)

**Table 2** Relative texture coefficients for (111), (200), (220), and (311) growth planes for each sample as measured with XRD

| Sample | RTC <sub>111</sub> | RTC <sub>200</sub> | RTC <sub>220</sub> | RTC <sub>311</sub> |
|--------|--------------------|--------------------|--------------------|--------------------|
| A1     | 5.26               | 4.85               | 72.32              | 17.66              |
| B1     | 8.09               | 4.32               | 71.84              | 15.74              |
| B2     | 7.80               | 4.27               | 72.66              | 15.26              |

# Application of Diffraction Data

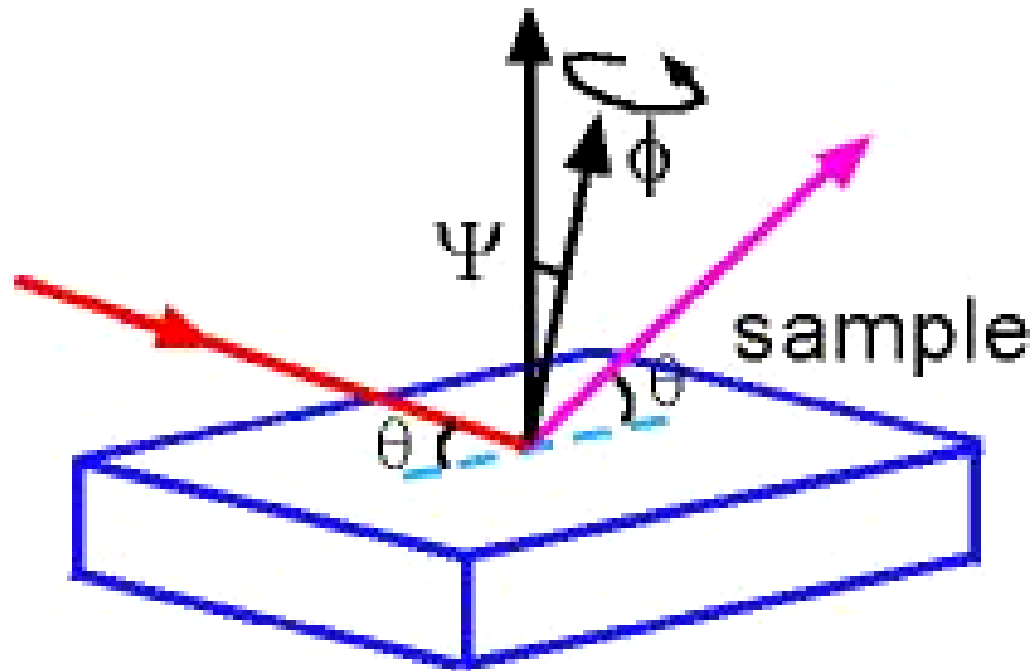
## Pole Figure Analysis

### Texture Analysis Experiment

Measurement of texture (the non-random or preferred orientation of crystallites) involves measurement of the variations in intensity of a single Bragg reflection as the sample is both tilted ( $\psi$ ) and rotated ( $\phi$ ).

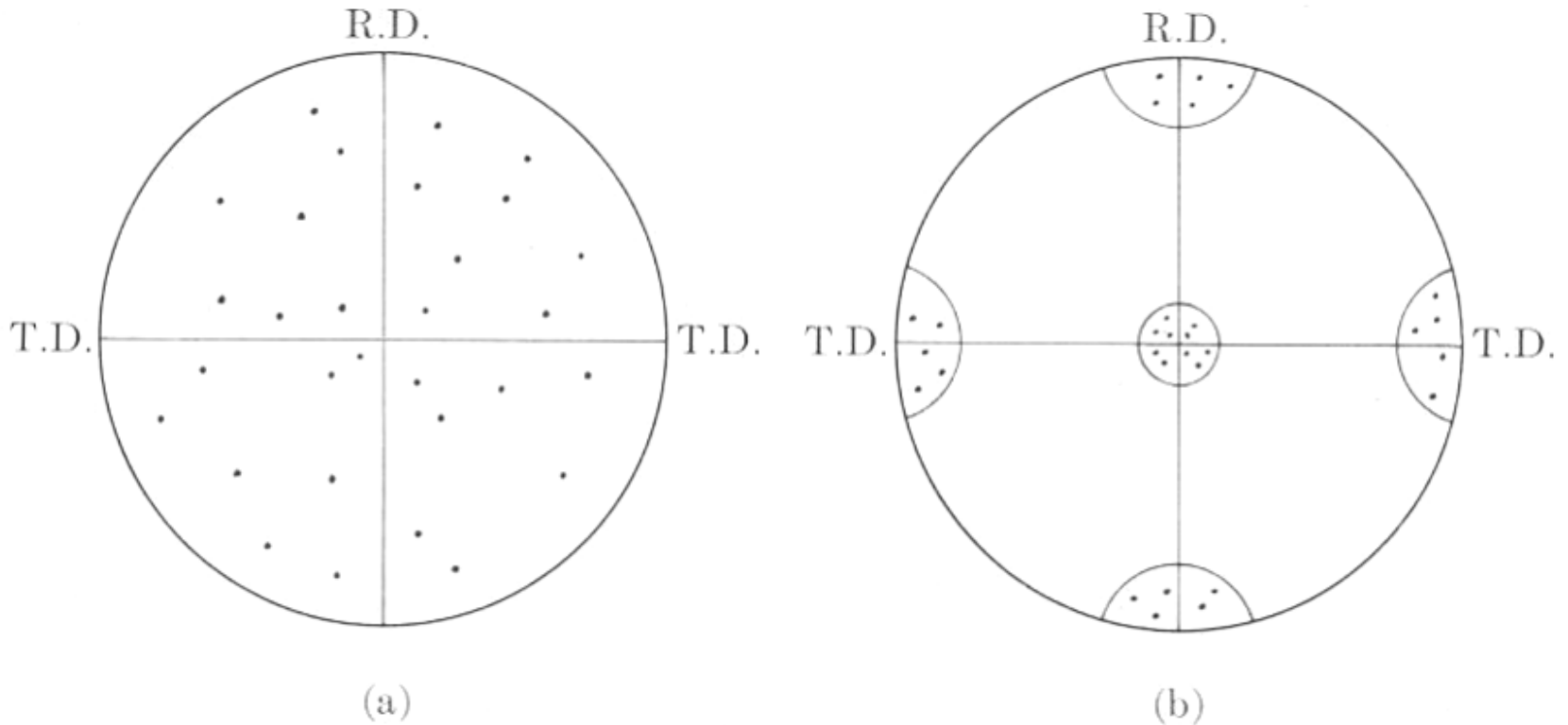
The result is plotted as a 'pole figure', in which the contours indicate intensity levels as a function of sample orientation.

# Application of Diffraction Data



## Pole Figure Measurement

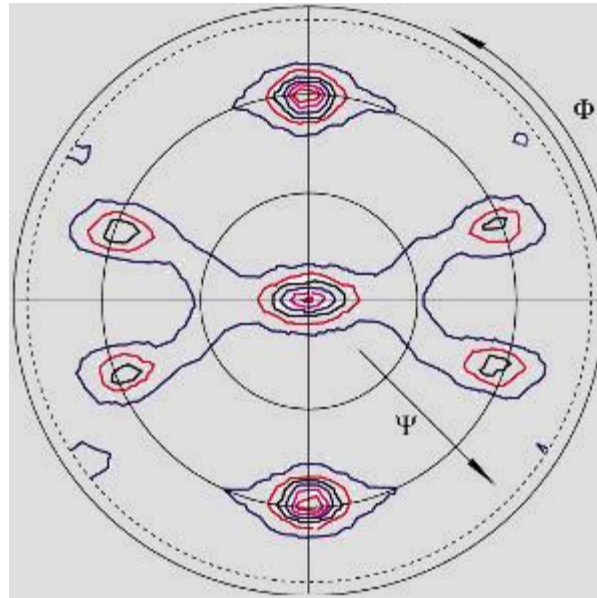
# Application of Diffraction Data



(100) pole figures for a sheet material  
(a) Random orientation (b) Preferred orientation

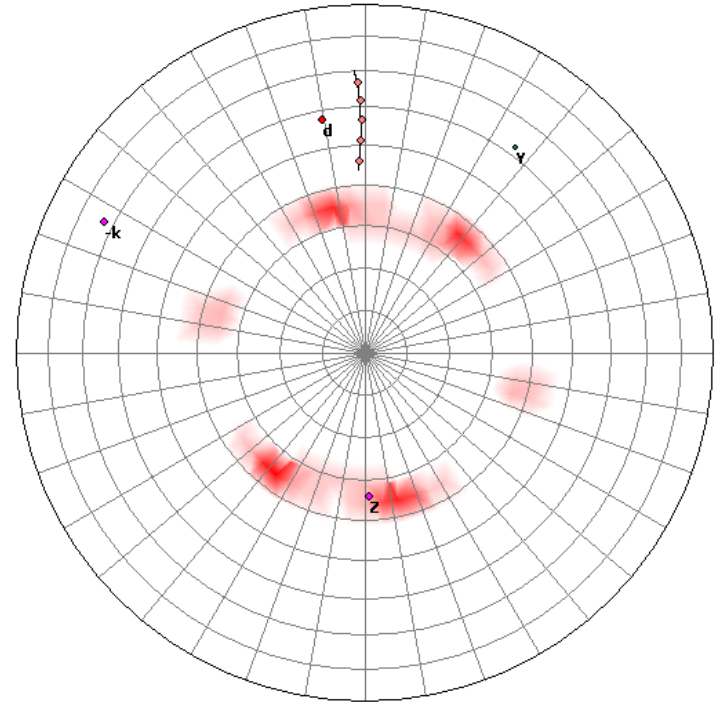
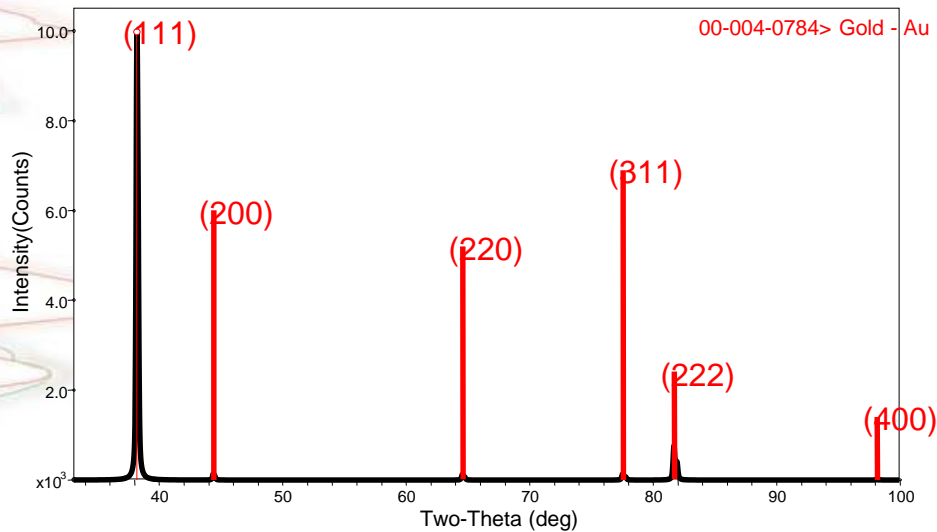
# Application of Diffraction Data

## Texture Analysis



# Application of Diffraction Data

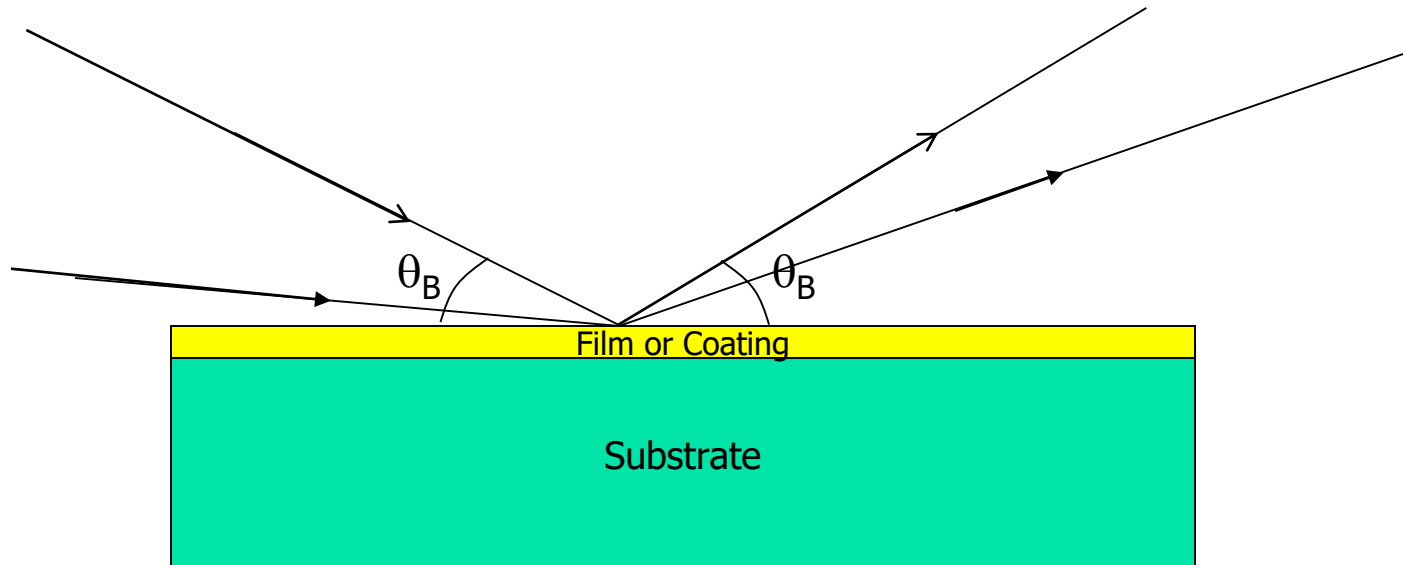
## Texture Analysis





# Application of Diffraction Data

## Thin Film Analysis



Example – thin films grown epitaxially on top of a substrate.

# Application of Diffraction Data

## Epitaxial Films

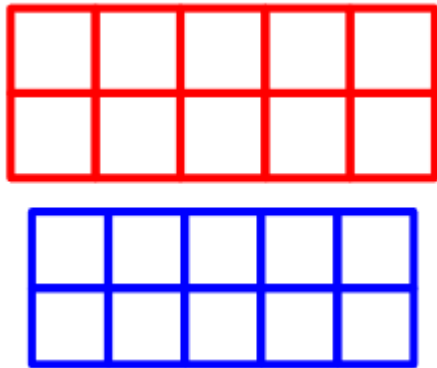
An epitaxial film is a lattice-matched thin film grown on usually a single crystal substrate

- The lattice of the film is nearly identical to the lattice of the substrate
- When the film grows, its lattice changes (strains) in order to match the lattice of the substrate
- The atomic bonding across the substrate and film is “perfectly” matched

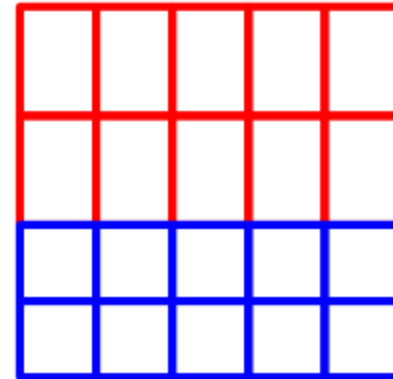
# Application of Diffraction Data

## Epitaxial Films

An epitaxial film is a lattice-matched thin film grown on usually a single crystal substrate



The lattice of the film (red) is almost the same as the substrate (blue)



The lattice of the epitaxial film (red) distorts to minimize the strain energy where it bonds to the substrate (blue)

# Application of Diffraction Data

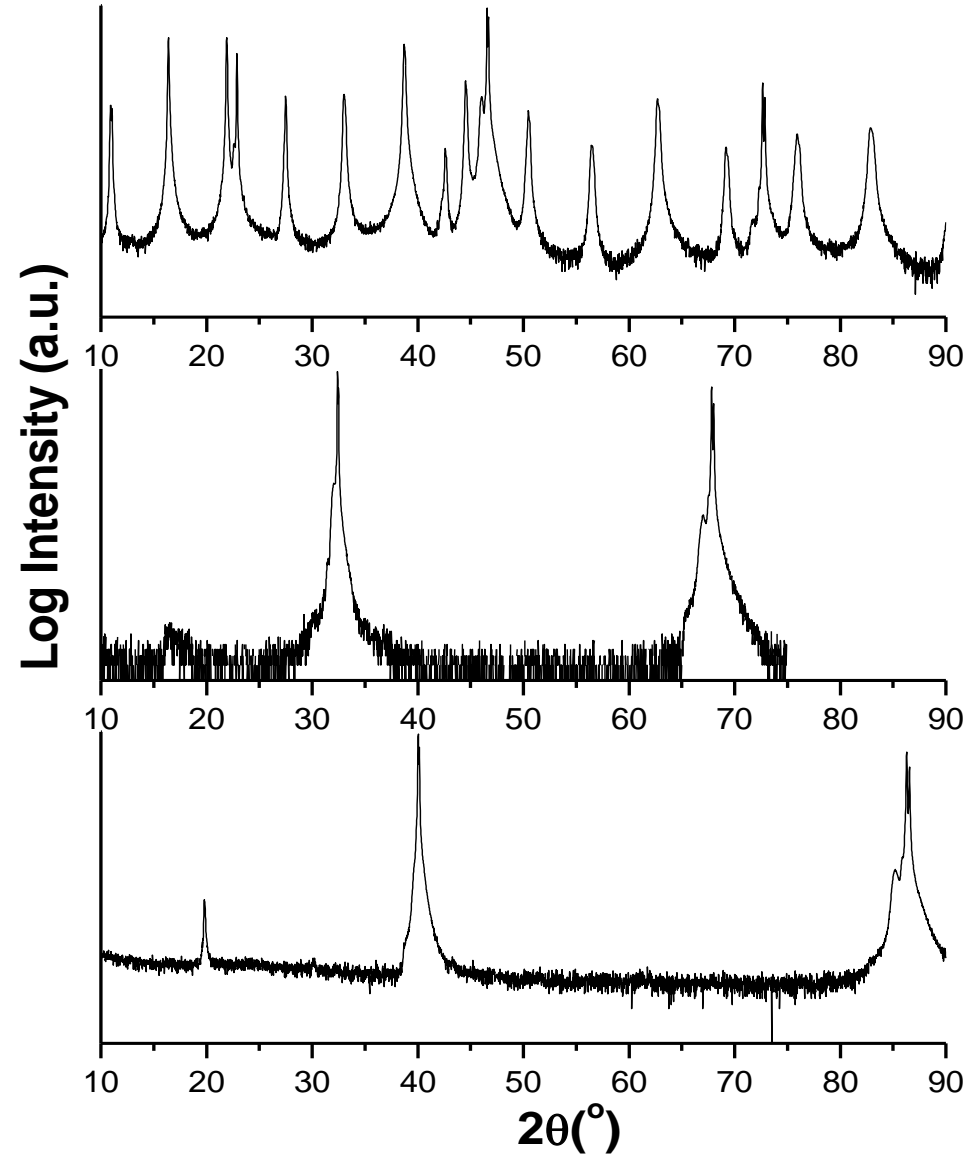
**Example:**

**Bismuth Titanate thin  
films on oriented SrTiO<sub>3</sub>  
substrates**

**Only one type of peaks**

**It apparent that films are  
highly oriented**

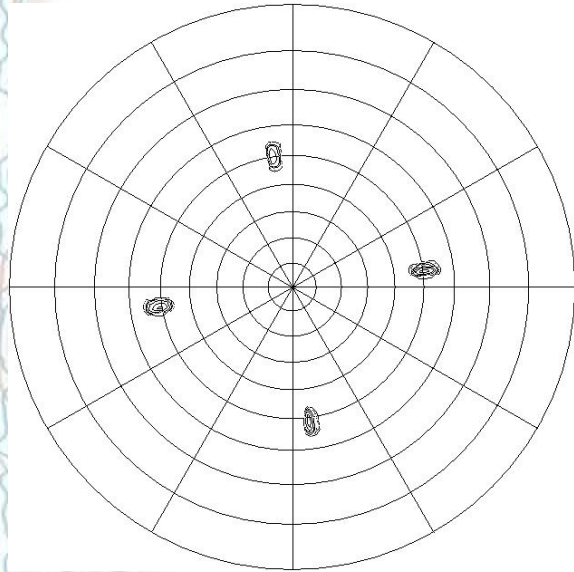
**3 different examples**



# Application of Diffraction Data

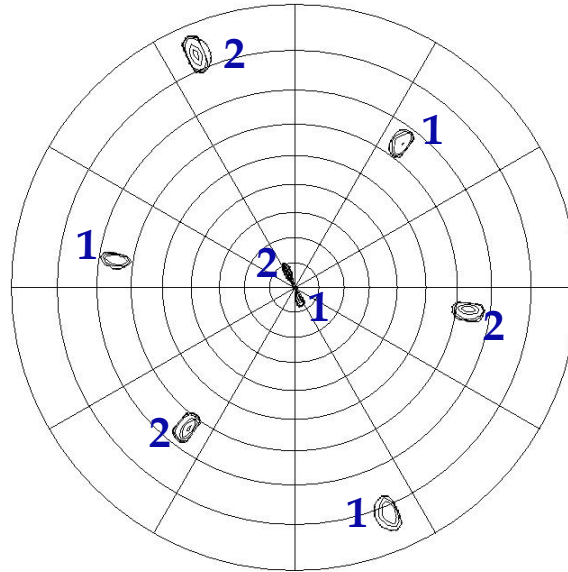
## Pole Figure Analysis

SrTiO<sub>3</sub> (100)



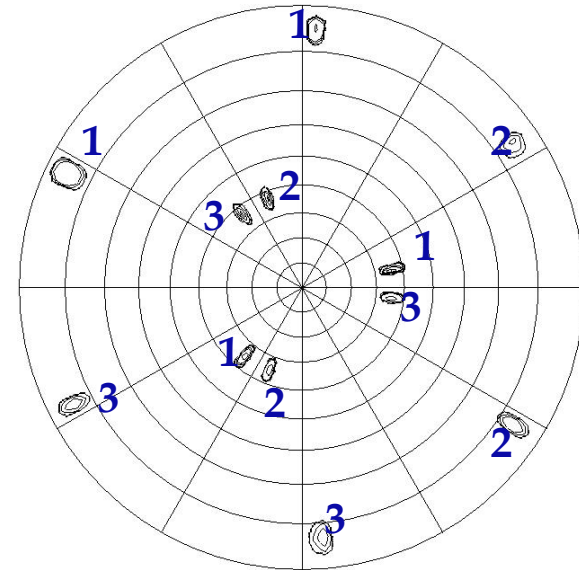
- 4 Peaks at ~50°
- Excellent in-plane orientation

SrTiO<sub>3</sub> (110)



- 2 sets of peaks at ~ 5, 65 and 85°
- Indicating a doublet or opposite twin growth

SrTiO<sub>3</sub> (111)

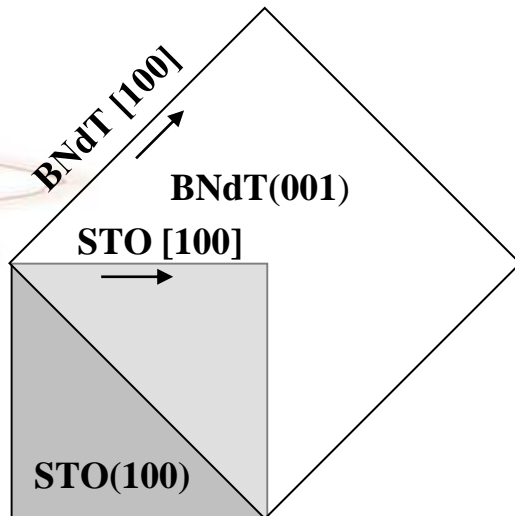
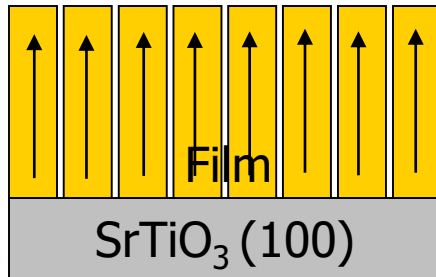


- 3 sets of peaks at ~ 35 and 85°
- indicating a triplet or triple twin growth



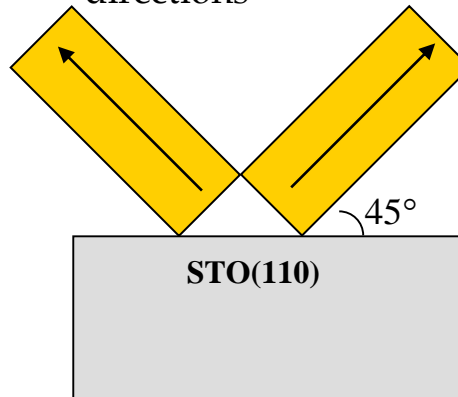
# Application of Diffraction Data

## Texture Analysis Evolution

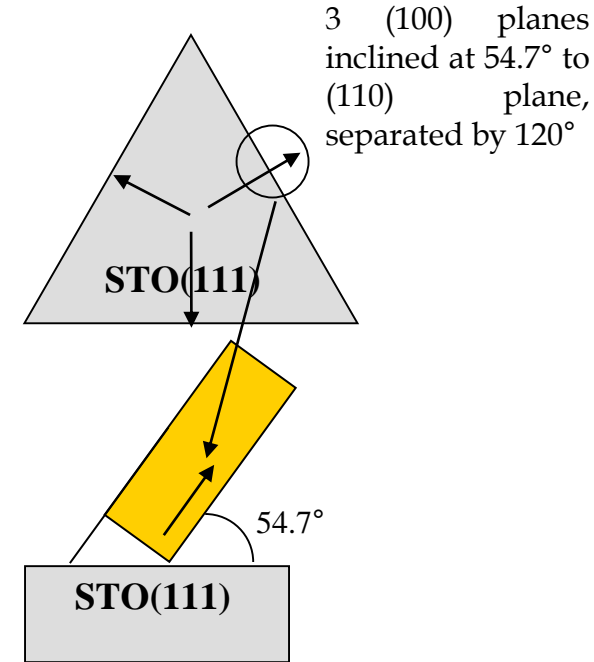


$\text{BNdT/SrTiO}_3 (100)$

Two  $(100)$  planes inclined at  $45^\circ$  to  $(110)$  plane in opposite directions



$\text{BNdT/SrTiO}_3 (110)$



$\text{BNdT/SrTiO}_3 (111)$

# Application of Diffraction Data

## Rocking Curves

A useful method for evaluating the quality of oriented samples such as epitaxial films.

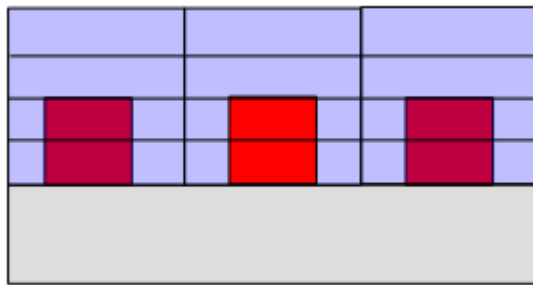
RC measurement reveals broadening of the diffraction peaks. The increase in the peak widths can be caused by:

- Mosaicity (misorientation of crystallites)
- Strain
- Limited layer thickness

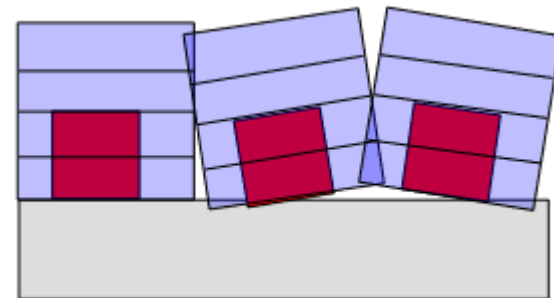
# Application of Diffraction Data

## Epitaxial Films – Mosaic Spread

**Mosaicity is created by slight misorientations of different crystals as they nucleate and grow on the substrate. When the crystals join, they form low energy domain boundaries.**



Ideal case, each nuclei (red) is perfectly oriented. When the crystals grow and meet, there is perfect bounding between the crystallites and therefore there is no grain boundary



If the nuclei (red) are slightly misaligned, then low angle domain boundaries will be formed.



# Application of Diffraction Data

## Rocking Curves

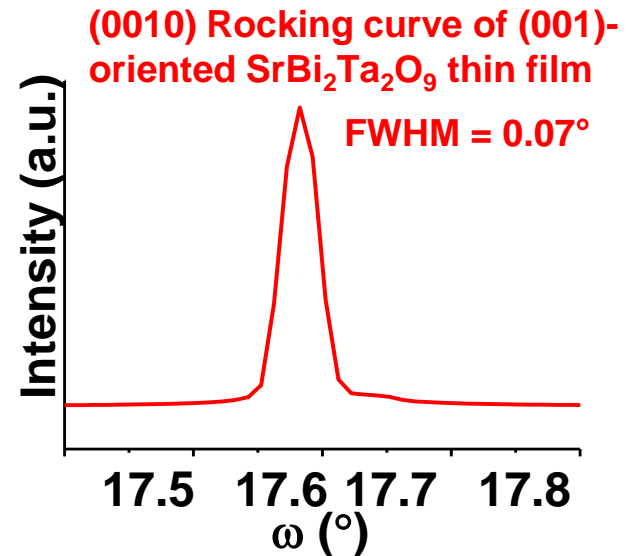
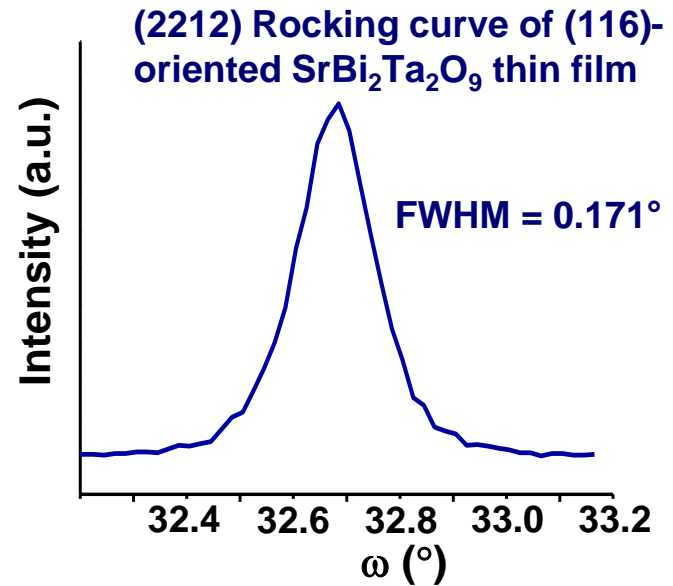
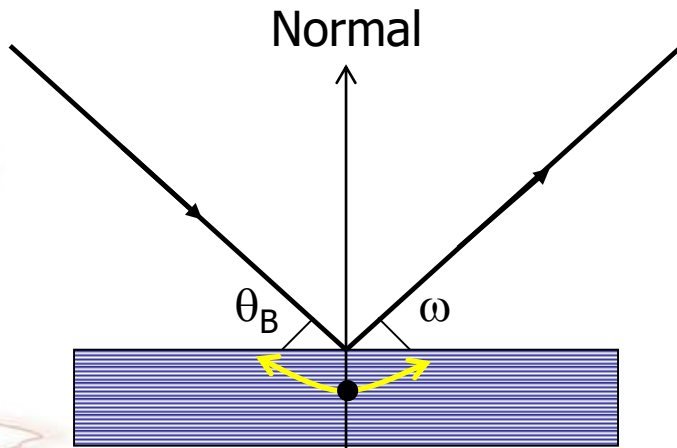
Typically XRD peak width is quantitatively described by Full-Width-at Half-Maximum (FWHM).

$\omega$  is changed by rocking the sample but  $\theta_B$  is held constant

Width of Rocking curve is a direct measure of the range of orientation present in the irradiated area of the crystal

# Application of Diffraction Data

## Rocking Curves



## Summary

|                                   |  |
|-----------------------------------|--|
| Perfect Epitaxy                   | Single crystal film in perfect registry with a substrate. There are no defects in the film or the substrate.   |
| Nearly perfect epitaxy            | Single crystal film in nearly perfect registry with a substrate. Both film and substrate contain a low concentration of defects. Most defects are dislocations in the film.                                |
| Textured epitaxial*               | Film consists of mosaic domains in nearly perfect registry with the substrate. All domain boundaries are very low angle/low energy. There is nearly perfect bonding across domain boundaries.              |
| Strongly textured polycrystalline | Film consists of grains with nearly perfect preferential orientation of all principle axes. This orientation is often strongly correlated to the substrate. Misorientation parameter for texture is small. |
| Textured polycrystalline          | Film consists of grains with a preferred orientation for 3 principle axes or only along 1 axis out-of-plane.   |
| Polycrystalline                   | Film consists of randomly oriented grains.   |
| Amorphous                         | Film does not have long-range crystalline order.   |

# Application of Diffraction Data

## Summary

|                                      | Thickness     | Composition                  | Lattice Strain/<br>Relaxation | Defects                     | Orientation    | Residual<br>Stress | Crystallite<br>Size |
|--------------------------------------|---------------|------------------------------|-------------------------------|-----------------------------|----------------|--------------------|---------------------|
| Perfect Epitaxy                      | XRR,<br>HRXRD | HRXRD, RC                    | Assume 100%                   | Assume<br>none              | HRXRD          | --                 | --                  |
| Nearly perfect<br>epitaxy            | XRR,<br>HRXRD | HRXRD, RC                    | HRXRD                         | RC                          | HRXRD          | --                 | --                  |
| Textured<br>epitaxial*               | XRR,<br>HRXRD | HRXRD                        | HRXRD, IP-<br>GIXD            | RC                          | HRXRD          | --                 | --                  |
| Strongly textured<br>polycrystalline | XRR           | XRPD, IP-<br>GIXD            | IP-GIXD                       | XRPD,<br>IP-GIXD            | IP-GIXD,<br>PF | IP-GIXD            | XRPD, IP-<br>GIXD   |
| Textured<br>polycrystalline          | XRR           | XRPD,<br>GIXD or IP-<br>GIXD | --                            | XRPD,<br>GIXD OR<br>IP-GIXD | PF             | Psi                | XRPD,<br>GIXD       |
| Polycrystalline                      | XRR           | XRPD,<br>GIXD                | --                            | XRPD,<br>GIXD               | PF             | Psi                | XRPD,<br>GIXD       |
| Amorphous                            | XRR           | --                           | --                            | --                          | --             | --                 | --                  |

**XRR**- X-Ray Reflectivity

**HRXRD**- High Resolution XRD using  
coupled scan or RSM

**RC**- Rocking Curve

**XRPD**- Bragg-Brentano powder diffraction

**GIXD**- grazing incidence XRD

**IP-GIXD**- in-plane grazing incidence XRD

**PF**- pole figure

**Psi**-  $\sin^2\psi$  using parallel beam

# Application of Diffraction Data

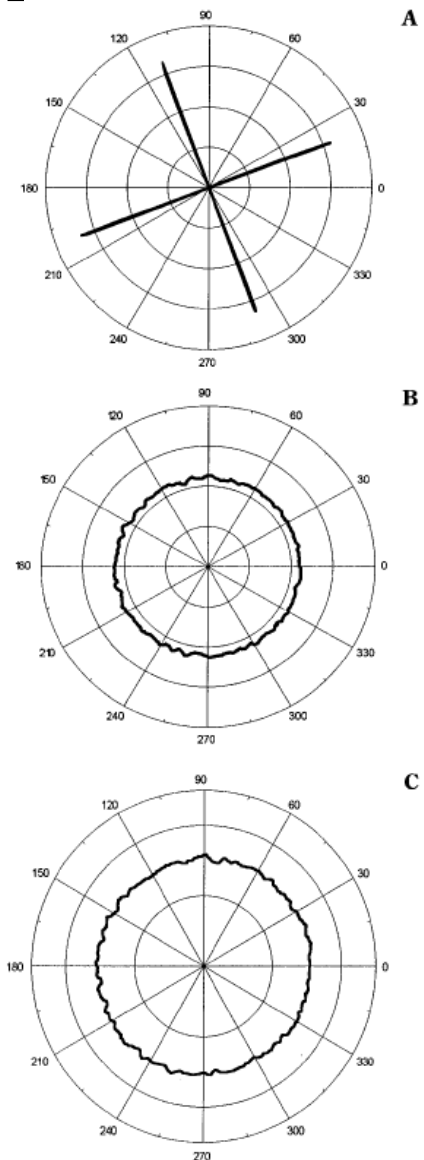


Figure 7. Rotational scans for (A) [100] silicon single crystal, (B) silicon powder, and (C) a 5  $\mu\text{m}$  thick [100]  $\text{Ti}_2\text{O}_3$  film. X-ray diffraction is from the {111} planes.

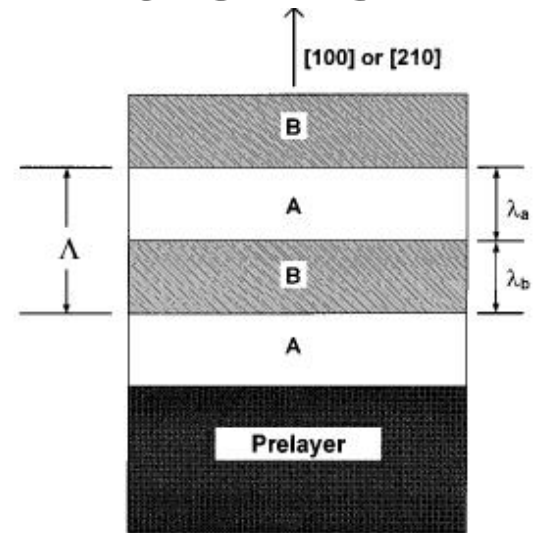


Figure 8. Idealized superlattice grown on an oriented prelayer. The orientation of the prelayer is maintained through the superlattice. The thickness of layer a is  $\lambda_a$ , and the thickness of layer b is  $\lambda_b$ .

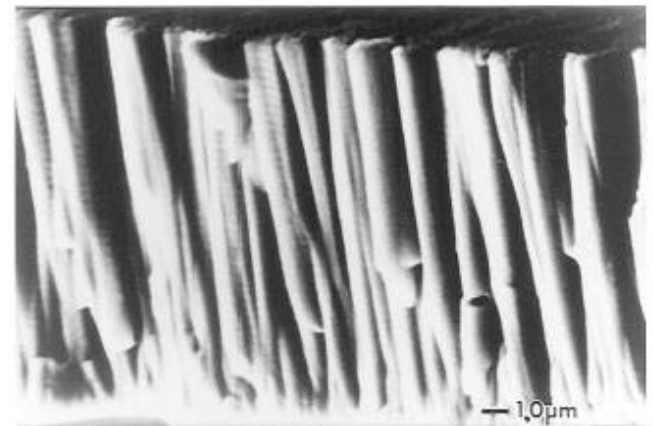
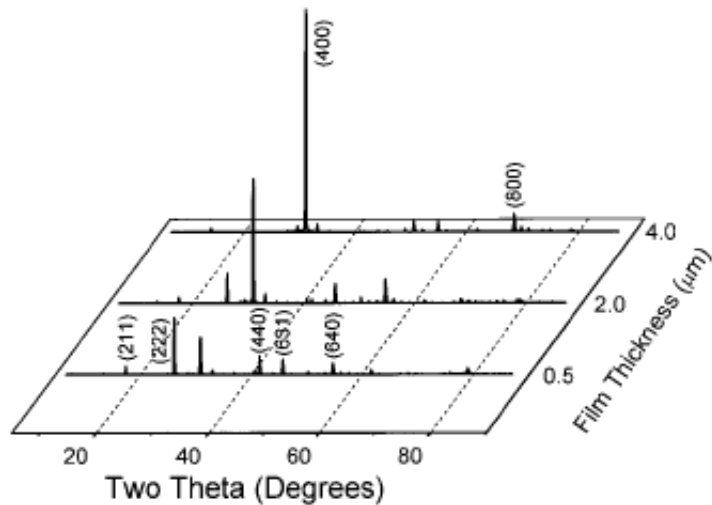
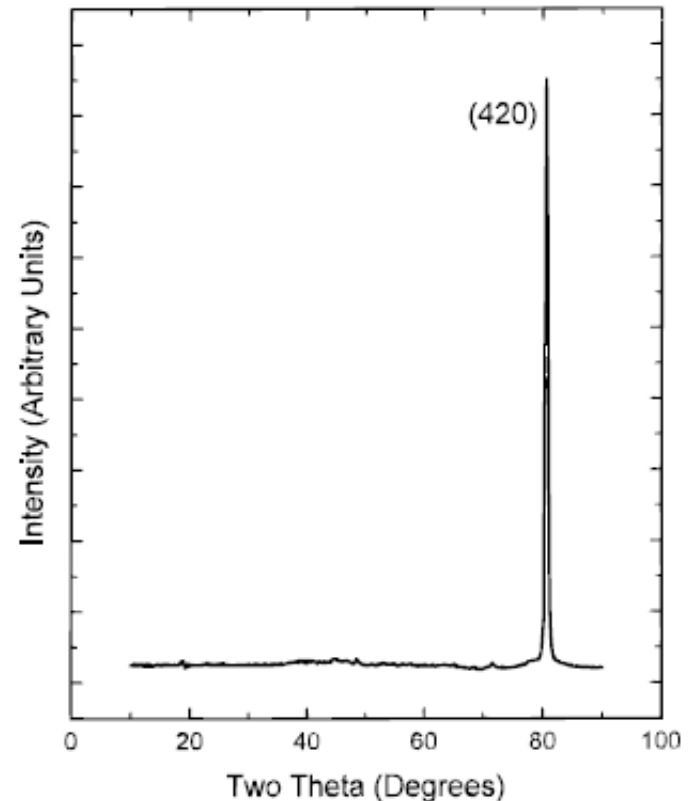


Figure 9. Scanning electron micrograph of an electrodeposited columnar superlattice structure. The individual layer thicknesses are approximately 0.5  $\mu\text{m}$ . The bar marker is 1.0  $\mu\text{m}$ .

# Application of Diffraction Data



**Figure 3.** Dependence of the prelayer texture on the film thickness for a  $\text{Ti}_2\text{O}_3$  prelayer. The texture increased with film thickness. A current density of  $5 \text{ mA/cm}^2$ , and a stirred solution of  $0.1 \text{ M TiNO}_3$  in  $5 \text{ M NaOH}$  was used to deposit the prelayers.



**Figure 4.** Highly oriented  $20 \mu\text{m}$   $\text{Pb-Ti-O}$  prelayer with a [210] texture. A current density of  $5 \text{ mA/cm}^2$ , and a stirred solution of  $0.005 \text{ M TiNO}_3$ ,  $0.1 \text{ M Pb(NO}_3)_2$  in  $5 \text{ M NaOH}$  was used to deposit the  $\text{Pb-Ti-O}$  film.

# Application of Diffraction Data

## XRD can be used for:

- Bravais lattice determination – phase determination (crystalline phases and orientation)
- Lattice parameter determination
- Determination of solvus line in phase diagrams (order-disorder transformation)
- Long range order (Texture Analysis)**
- Crystallite size and Strain
- Temperature factor – thermal diffuse scattering (thermal expansion)
- Thickness measurements of thin films and multilayers**

# Application of Diffraction Data

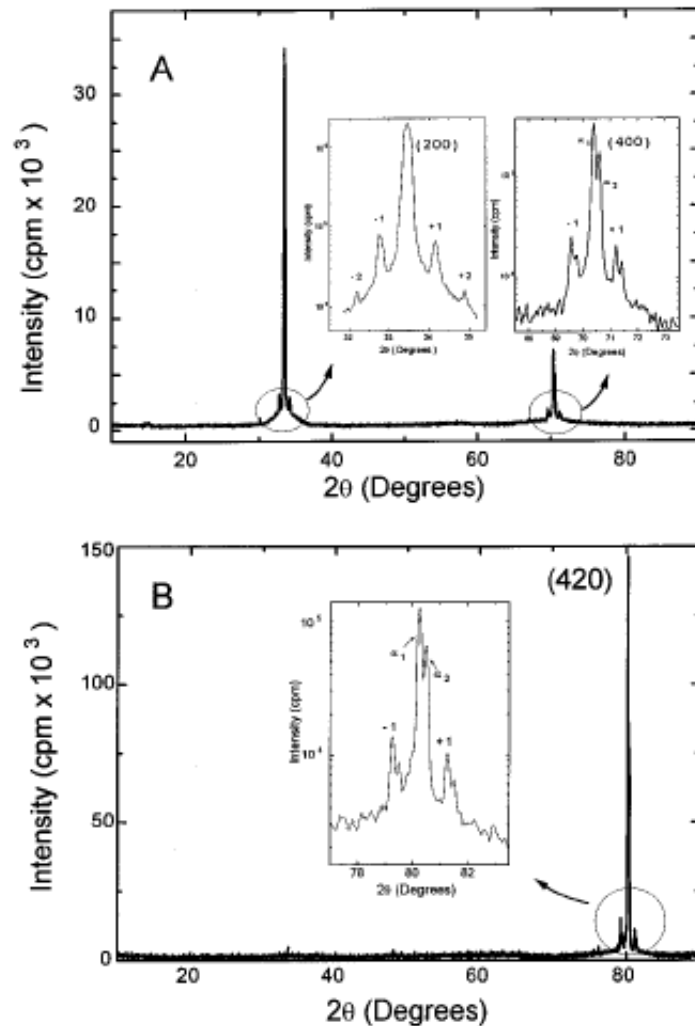


Figure 12. (A) [100]-textured superlattice in the Pb-Tl-O system. The modulation wavelength was calculated as  $\Lambda_F = 13.2$  nm by Faraday's law and measured by X-ray diffraction as  $\Lambda_x = 13.3$  nm. (B) [210]-textured superlattice with  $\Lambda_F = 12$  nm and  $\Lambda_x = 11.8$  nm. Both superlattices were deposited by pulsing the current between 0.05 mA/cm<sup>2</sup> (82 s) and 5 mA/cm<sup>2</sup> (1.2 s).

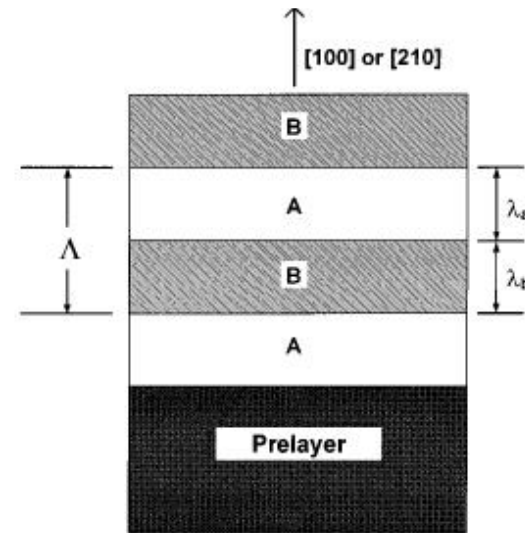
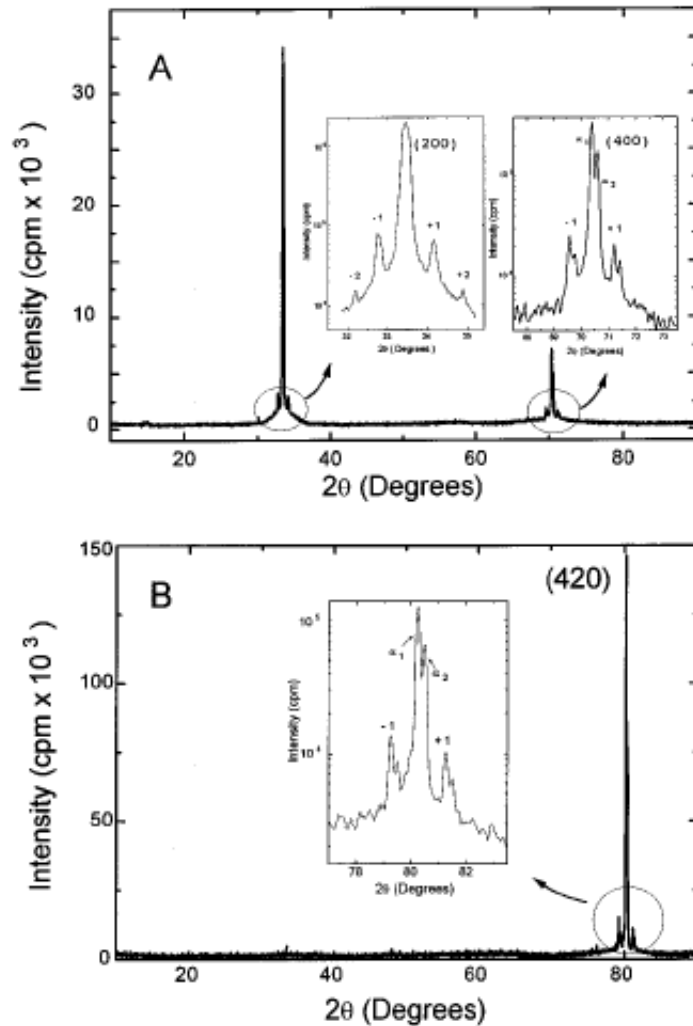


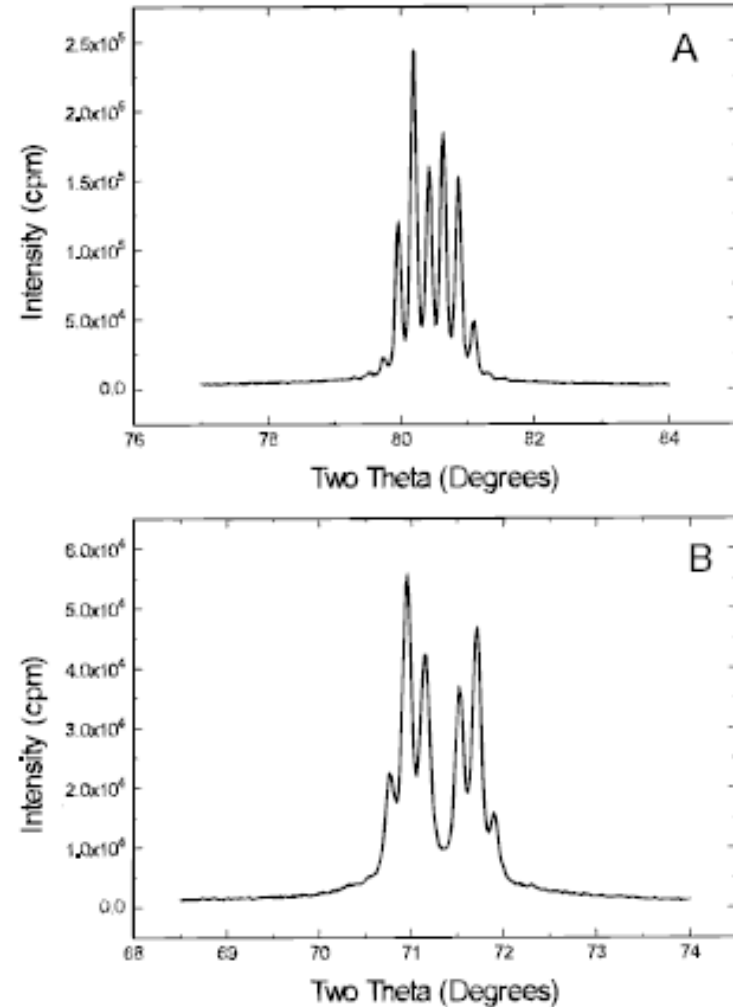
Figure 8. Idealized superlattice grown on an oriented prelayer. The orientation of the prelayer is maintained through the superlattice. The thickness of layer a is  $\lambda_a$ , and the thickness of layer b is  $\lambda_b$ .



# Application of Diffraction Data

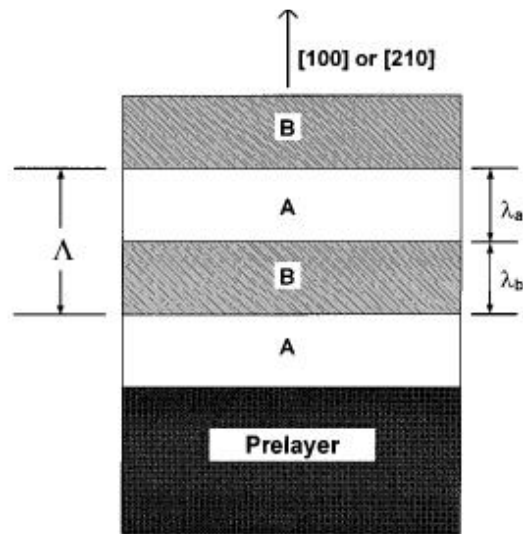


**Figure 12.** (A) [100]-textured superlattice in the Pb-Tl-O system. The modulation wavelength was calculated as  $\Lambda_F = 13.2$  nm by Faraday's law and measured by X-ray diffraction as  $\Lambda_x = 13.3$  nm. (B) [210]-textured superlattice with  $\Lambda_F = 12$  nm and  $\Lambda_x = 11.8$  nm. Both superlattices were deposited by pulsing the current between 0.05 mA/cm<sup>2</sup> (82 s) and 5 mA/cm<sup>2</sup> (1.2 s).

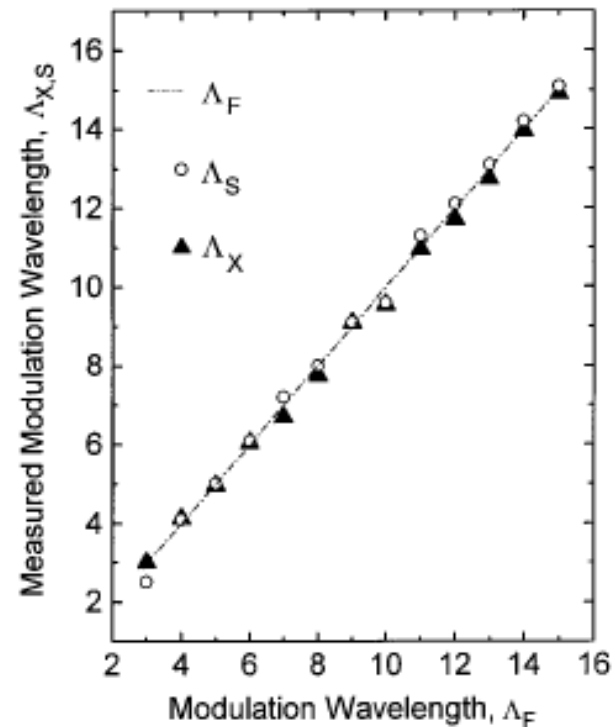


**Figure 14.** (A) X-ray diffraction pattern of a [210]-textured superlattice in the Pb-Tl-O system using Cu K $\alpha$  radiation as the X-ray source. The modulation wavelength was calculated as  $\Lambda_F = 50$  nm by Faraday's law. (B) Same as (A), except using Cu K $\beta$  radiation for the X-ray source.

# Application of Diffraction Data



**Figure 8.** Idealized superlattice grown on an oriented prelayer. The orientation of the prelayer is maintained through the superlattice. The thickness of layer a is  $\lambda_a$ , and the thickness of layer b is  $\lambda_b$ .



**Figure 13.** Comparison of the Pb–Ti–O superlattice modulation wavelength in nanometers calculated by Faraday's law ( $\Delta_F$ ) with those measured by scanning tunneling microscopy ( $\Delta_S$ ) and X-ray diffraction ( $\Delta_X$ ).

# **Homework Assignment: Phase Diagram Calculations**

**Due Tuesday 11-19-24**

**Read Chapter 8 from:**

**-X-ray Diffraction Procedures by Klug and Alexander**

**Read Chapter 3-7, 9-11 and 13 from:**

**-Introduction to X-ray powder  
Diffractometry by Jenkins and Synder**

**Read Chapter 3, 4, 6, 13, and 14 from**

**-Elements of X-ray Diffraction  
by Cullity and Stock**

**Read Chapter 2 from Norton**